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Synthesis of (–)-9,10-*epi*-Stemoamide

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Abstract: An efficient synthesis of (–)-9,10-*epi*-stemoamide has been accomplished in nine steps and 13% overall yield. The synthesis features a lithium hydroxide-promoted fragmentation and an intramolecular 7-*exo-trig* radical cyclization.

Extracts from the roots of Stemona tuberosa Lour. and related Stemona species have been used in traditional Chinese medicine as anticough agents and insecticides. In 1992, Xu and his colleagues isolated and elucidated the structures of several Stemona alkaloids.¹ (-)-Stemoamide (1), one of these Stemona alkaloids, is composed of a perhydropyrroloazepine ring fused to a γ -butyrolactone moiety, and it contains four contiguous stereogenic centers. Due to its interesting biological activity and structural complexity, it has been a challenging target for synthetic organic chemists. To date, there have been five total syntheses of stemoamides reported in the literature. In 1994, Williams et al. reported the first total synthesis of (-)-stemoamide by a linear approach starting from (*R*)-methyl 3-hydroxy-2-methylpropionate.^{2a} The syntheses of both (\pm) - and (-)-stemoamides by Mori et al. featured a ruthenium-catalyzed enyne metathesis as a key step.^{2b,c} Narasaka et al. utilized an oxidative coupling reaction of α -stannyl pyrrolidinone with silyl enol ether.^{2d} The total syntheses by Jacobi et al. of both (\pm) - and (-)-stemoamides used alkyne oxazole Diels-Alder and retro-Diels-Alder reaction sequences to establish the tricyclic skeleton of stemoamide.^{2e,f} Recently, Gurjar et al. reported a formal synthesis utilizing a ringclosing metathesis.^{2g} Herein, we report an efficient synthesis of (-)-9,10-epi-stemoamide (2) via a sequential asymmetric Birch reduction-alkylation, lithium hydroxide-promoted fragmentation, and a 7-exo-trig radical reaction.



The synthesis commenced with the preparation of the intermediate 7, as shown in Scheme 1. Following the



 a Reagents and conditions: (a) K, NH₃, THF, *t*-BuOH (1 equiv), -78 °C, piperylene, MeI; (b) 6 N HCl, MeOH, rt; (c) I₂, THF/H₂O (1:1), rt, 81% for three steps; (d) LiOH·H₂O, THF/H₂O (10:1), rt, 41% for 7 and 40% for **8**.

procedure developed in our laboratory, Birch reduction of **3** under the standard reaction conditions (K, NH₃, THF, *t*-BuOH, -78 °C) provided the chiral enolate, which was then alkylated in situ with methyl iodide to yield the corresponding 1,4-cyclohexadiene **4** as a single diastereomer.³ Enol ether hydrolysis of **4** with 6 N aqueous HCl solution in MeOH at room temperature gave the β , γ enone **5**. Subsequent iodolactonization of **5** with I₂ in THF/H₂O (1:1) at room temperature afforded the enantiomerically pure carbolactone **6** in 81% isolated yield for three steps. Treatment of **6** with lithium hydroxide monohydrate in THF/H₂O (10:1) at room temperature provided the requisite intermediate, butenolide carboxylic acid **7** in 41% yield, along with the side product, 2-methyl-4-hydroxy-2-cyclohexen-1-one (**8**) in 40% yield.⁴⁻⁶

Next, the butenolide carboxylic acid **7** was converted into the phenylthiolactam **12** for radical cyclization, as depicted in Scheme 2. First, **7** was reduced uneventfully to provide the corresponding alcohol **9** in 92% yield by using a borane-tetrahydrofuran complex (BH₃·THF) in THF at -30 °C.⁷ Mitsunobu reaction⁸(Ph₃P, DEAD, THF, 25 °C) of **9** with succinimide afforded the cyclic imide **10**

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SCHEME 2^a



^{*a*} Reagents and conditions: (a) BH₃·THF, THF, -30 °C, 92%; (b) succinimide, Ph₃P, DEAD, THF, rt, 92%; (c) NaBH₄, MeOH, -10 °C, 88%; (d) PhSH, TsOH·H₂O, benzene, 0 °C, 70%; (e) *n*-Bu₃SnH, AIBN, benzene, reflux, 24 h, [c] = 3.5 mM; (f) K₂CO₃, MeOH, rt, 74% for two steps.

in 92% yield. Reduction of cyclic imide was effected with NaBH₄ in MeOH at -10 °C to provide the carbinol lactam **11** in 88% yield.⁹ Direct treatment of **11** with benzenethiol and *p*-toluenesulfonic acid monohydrate in benzene at 0 °C afforded the required radical reaction substrate, phenylthiolactam **12** in 70% yield.¹⁰

With phenylthiolactam 12 in hand, we executed the key step, a radical-promoted cyclization.¹¹ Thus, employing the standard radical reaction protocol, a solution of tributyltin hydride and 2,2'-azobisisobutyronitrile (AIBN) in benzene was added in a period of 24 h to the gently refluxing solution of phenylthiolactam 12 in benzene (final concentration \sim 3.5 mM). To our delight, the desired intramolecular 7-exo-trig radical reaction¹² took place in a remarkably efficient manner, resulting in the formation of a mixture of two diastereomers of constitutional stemoamide (2 and 13) as an inseparable mixture $(\sim 10:1)$.¹³ As expected, compounds **2** and **13** are epimers at C-10 with a cis ring fusion stereorelationship at C-8 and C-9 and resulted from the complete facial selective addition (β -face addition) of an amido radical to a conjugated C–C double bond of the γ -lactone moiety. Subjection of these two diastereomers to K₂CO₃ in MeOH at room temperature furnished only a single stereoisomer **2** in 74% isolated yield. The facile epimerization at C-10 of 13 to 2 is in good accord with the fact that α -Me isomer

2 is thermodynamically more stable than β -Me isomer **13** by 2.8 kcal/mol (MM2 calculations).¹⁴

In summary, we have demonstrated an efficient synthesis of (-)-9,10-*epi*-stemoamide **2**. Lithium hydroxidepromoted fragmentation of carbolactone to butenolide carboxylic acid (**6** \rightarrow **7**) and the intramolecular 7-*exo-trig* radical cyclization reaction of phenylthiolactam to (-)-9,10-*epi*-stemoamide (**12** \rightarrow **2**) are noteworthy.

Experimental Section¹⁵

(5R)-5-(3'-Hydroxypropyl)-3-methyl-2(5H)-furanone (9). To a solution of butenolide acid 7 (250 mg, 1.5 mmol) in THF (20 mL) was added dropwise BH3·THF (1.0 M in THF, 1.6 mL, 1.6 mmol) at 0 °C. The resulting solution was stirred overnight at room temperature. The reaction mixture was treated with H₂O and extracted with CH₂Cl₂. The organic layers were isolated, dried over anhydrous Na₂SO₄, and concentrated in vacuo. Flash column chromatography (hexanes/EtOAc, 1:1) of the crude residue gave **9** (210 mg, 92%) as a colorless oil: $[\alpha]^{23}_{D}$ -37.0 (c 1.27, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ 7.04-7.03 (m, 1H), 4.91-4.88 (m, 1H), 3.62-3.58 (m, 2H), 2.76 (br s, 1H), 1.84 (d, J = 1.9 Hz, 3H), 1.98–1.79 (m, 1H), 1.67–1.57 (m, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 174.5 (s), 149.0 (d), 129.6 (s), 80.9 (d), 61.6 (t), 29.7 (t), 27.8 (t), 10.4 (q); IR ν 3422, 2930, 1743 cm⁻¹; CIMS 157 (M⁺ + 1, 100); HRMS calcd for $C_8H_{13}O_3$ (M⁺ + 1) 157.0870, found 157.0865.

Cyclic Imide (10). To a solution of alcohol 9 (270 mg, 1.7 mmol), triphenylphosphine (500 mg, 1.9 mmol), and succinimide (171 mg, 1.7 mmol) in THF (5 mL) was added diethyl azodicarboxylate (DEAD, 300 μ L, 1.9 mmol). The resulting solution was stirred overnight and then concentrated in vacuo. Diethyl ether was added to the residue, and after removal of a white precipitate, the solution was concentrated in vacuo. The crude residue was purified by flash column chromatography (hexanes/ EtOAc, 1:2) to give the cyclic imide 10 (377 mg contaminated with a small amount of triphenylphosphine oxide, 92%) as a colorless oil: [α]²⁷_D -29.7 (c 0.74, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ 6.96 (quint, J = 1.5 Hz, 1H), 4.84 (m, 1H), 3.50–3.44 (m, 2H), 2.65 (s, 4H), 1.83 (d, J = 1.5 Hz, 3H), 1.74–1.59 (m, 3H), 1.54–1.48 (m, 1H); ¹³C NMR (126 MHz, CDCl₃) δ 177.2, 173.9, 148.1, 130.3, 80.2, 38.1, 30.6, 28.1, 23.4, 10.5; IR (neat) ν 2944, 1752, 1700 cm⁻¹; CIMS 238 (M⁺ + 1, 100); HRMS calcd for $C_{12}H_{16}NO_4$ (M⁺ + 1) 238.1079, found 238.1080.

Carbinol Lactam (11). To a solution of cyclic imide 10 (408 mg, 1.7 mmol) in MeOH (5 mL) was added portionwise NaBH₄ (650 mg, 17 mmol) at -10 °C. The resulting solution was stirred for 0.5 h at the same temperature and quenched with water (10 mL). The reaction mixture was extracted with CH_2Cl_2 (5 \times 10 mL). The combined organic layers were dried over anhydrous Na₂SO₄ and concentrated in vacuo to give the residue (388 mg, 94%). Flash column chromatography (EtOAc) of the residue gave the carbinol lactam **11** (362 mg, 88%) as a colorless oil: $[\alpha]^{25}$ _D -29.4 (c 0.34, CHCl₃); ¹H NMR (diastereomeric mixture, 500 MHz, CDCl₃) δ 7.04 (q, J = 1.5 Hz) and 7.03 (q, J = 1.5 Hz, 1H), 5.23-5.02 (m, 1H), 4.96-4.91 (m, 1H), 3.67-3.04 (m, 2H), 2.58-2.50 (m, 1H), 2.37–2.22 (m, 3H), 1.85 (d, J = 1.7 Hz) and 1.85 (d, J = 1.9 Hz, 3H), 2.02–1.45 (m, 5H); ¹³C NMR (diastereomeric mixture, 126 MHz, CDCl₃) δ 175.1, 175.0, 174.5 (overlapped), 149.0, 148.9, 129.9, 129.8, 83.4, 83.1, 80.9, 80.7, 39.7, 39.3, 30.6, 30.5, 28.9, 23.4, 23.3, 10.5; IR (neat) v 3403, 2930, 1752, 1670 cm⁻¹; CIMS 222 (M⁺ - 18, 100); HRMS calcd for C₁₂H₁₈NO₄ (M⁺ + 1) 240.1235, found 240.1226.

Phenylthiolactam (12). To a solution of carbinol lactam **11** (117 mg, 0.50 mmol) in benzene (3 mL) was added *p*-TsOH (1 mg) and thiophenol (75 μ L, 0.73 mmol) at 0 °C. The resulting solution was stirred at room temperature overnight, diluted with diethyl ether, and washed with water. The organic layers were

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⁽¹⁰⁾ For the preparation of phenylthiolactam and its use in radical cyclization reactions, see: Choi, J.-K.; Hart, D. J. *Tetrahedron* **1985**, *41*, 3959 and references therein.

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⁽¹²⁾ For a few examples of seven-membered ring formations by intramolecular radical cyclizations, see: (a) Beckwith, A. L. J.; Moad, G. J. Chem. Soc., Chem. Commun. **1974**, 472. (b) Beckwith, A. L. J.; Roberts, D. H. J. Am. Chem. Soc. **1986**, 108, 5893. (c) Bachi, M. D.; Hoornaert, C. Tetrahedron Lett. **1981**, 22, 2693. (d) Ghosh, A. K.; Ghosh, K.; Pal, S.; Ghatak, U. R. J. Chem. Soc., Chem. Commun. **1993**, 809. (e) Rigby, J. M.; Qabar, M. N. J. Org. Chem. **1993**, 58, 4473. (f) Rigby, J. M.; Laurent, S.; Cavezza, A.; Heeg, M. J. J. Org. Chem. **1998**, 63, 5587.

⁽¹³⁾ The ratio was determined by $^1\mathrm{H}$ NMR analysis of the crude reaction mixture.

⁽¹⁴⁾ $^{1}\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectra of $\boldsymbol{2}$ were identical to those prepared by Jacobi and Lee (see ref 2e).

⁽¹⁵⁾ For general experimental procedures and full characterization data of 3-8, see ref 4b.

isolated, dried over anhydrous Na₂SO₄, and concentrated in vacuo. The residue was purified by flash column chromatography (hexanes/EtOAc, 1:3) to give the phenylthiolactam 12 (112 mg, 70%) as a colorless oil: $[\alpha]^{25}_{D} - 10.2$ (c 0.59, CHCl₃); ¹H NMR (diastereomeric mixture, 500 MHz, CDCl₃) δ 7.41–7.39 (m, 2H), 7.35–7.30 (m, 3H), 7.00 (br t, J = 1.5 Hz, 1H), 6.98 (br t, J =1.5 Hz, 1H), 4.92-4.87 (m, 2H), 3.80-3.69 (m, 1H), 3.35-3.27 (m, 1H), 2.49-2.40 (m, 1H), 2.22-2.17 (m, 1H), 2.12-2.05 (m, 1H), 1.90 (br t, J = 1.5 Hz, 3H), 1.89 (br t, J = 1.5 Hz, 3H), 1.81-1.62 (m, 3H), 1.60-1.46 (m, 2H); ¹³C NMR (diastereomeric mixture, 126 MHz, CDCl₃) & 174.7, 174.6, 174.1, 174.0, 148.5, 148.3, 135.0, 134.9, 132.1, 130.6, 130.5, 130.2, 130.1, 129.3, 129.0, 128.9, 80.3, 80.2, 67.1, 67.0, 39.7, 39.4, 30.7, 30.6, 29.1, 29.0, 26.6, 26.5, 22.7, 22.5, 10.6, 10.5; IR (neat) ν 1752, 1686 cm⁻¹; CIMS 332 (M⁺ + 1, 3), 222 (100); HRMS calcd for C₁₈H₂₂NO₃S $(M^+ + 1)$ 332.1320, found 332.1315.

(-)-9,10-epi-Stemoamide (2). To a solution of phenylthiolactam 12 (74 mg, 22 μ mol, 5.5 mM) in benzene (40 mL) was added a solution of n-Bu₃SnH (93 µL, 33 mmol) and AIBN (11 mg, 6.7 μ mol) in benzene (25 mL) via a syringe pump over a period of 24 h at 85 °C (oil bath temperature). The resulting solution was stirred at reflux for an additional 24 h. The reaction mixture was cooled to room temperature and concentrated in vacuo. The residue was dissolved in diethyl ether, and DBU16 was added until the white precipitate remained. Then, iodine was added until the yellow color remained. The resulting suspension was passed through a short column to remove tin material. The solution was concentrated in vacuo, dissolved in MeOH (2 mL), and treated with K_2CO_3 (30 mg, 22 μ mol). The resulting reaction mixture was stirred at room temperature for 48 h and concentrated. The residue was then treated with water, neutralized with 10% aqueous HCl solution, and extracted with

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CH₂Cl₂. The organic layers were isolated, dried over anhydrous Na₂SO₄, and concentrated in vacuo. The residue was purified by flash column chromatography (hexanes/EtOAc, 1:10) to give 2 (37 mg, 74%) as the only detectable product, a colorless semisolid: $[\alpha]^{21}D$ -63.3 (c 1.69, MeOH); ¹H NMR (500 MHz, CDCl₃) δ 4.60 (ddd, J = 10.6, 7.8, 2.7 Hz, 1H), 4.17 (dt, J = 13.8, 4.7 Hz, 1H), 3.61 (dd, J = 9.8, 7.8 Hz, 1H), 2.76 (ddd, J = 13.9, 10.6, 3.4 Hz, 1H), 2.58-2.53 (m, 1H), 2.48 (m, 1H), 2.41 (m, 1H), 2.35-2.24 (m, 2H), 2.11-2.06 (m, 1H), 1.95-1.80 (m, 3H), 1.65-1.57 (m, 1H), 1.37 (d, J = 7.1 Hz, 3H); ¹H NMR (500 MHz, C₆D₆) δ 3.98 (dtd, J = 13.6, 4.6, 0.8 Hz, 1H, H-5a), 3.80 (ddd, J = 11.0, 8.1, 3.0 Hz, 1H, H-8), 2.42 (ddd, J = 9.9, 8.1, 1.6 Hz, 1H, H-9a), 2.07 (ddd, J = 13.7, 10.5, 3.4 Hz, 1H, H-5b), 1.95 (ddd, J = 16.6, 8.5, 10.9 Hz, 1H, H-2), 1.82 (ddd, J = 16.6, 9.5, 1.1 Hz, 1H, H-2), 1.65 (dq, J = 8.6, 7.4 Hz, 1H, H-10), 1.48-1.44 (m, 1H, H-7), 1.37 (dt, J = 10.1, 8.3 Hz, 1H, H-9), 1.28–1.20 (m, 1H, H-1b), 1.19-1.08 (m, 2H, H-6a and H-7), 1.00-0.92 (m, 1H, H-6a), 0.84 (d, J = 7.4 Hz, 3H, H-12), 0.85–0.80 (m, 1H, H-1a); ¹³C NMR (126 MHz, CDCl₃) & 177.9, 174.6, 80.7, 60.5, 50.7, 44.0, 39.0, 30.0, 28.9, 25.4, 23.9, 15.8; 13 C NMR (126 MHz, C₆D₆) δ 177.4 (s), 173.6 (s), 80.4 (d), 60.0 (d), 50.6 (d), 43.9 (t), 38.9 (d), 30.3 (t), 29.1 (t), 25.5 (t), 24.6 (t), 15.9 (q); IR (neat) v 2938, 1769, 1686 cm⁻¹; CIMS 224 (M⁺ + 1, 100); HRMS calcd for $C_{12}H_{18}$ - NO_3 (M⁺ + 1) 224.1286, found 224.1289.

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Supporting Information Available: ¹H and ¹³C NMR spectra of new compounds **2** and **9–12**. This material is available free of charge via the Internet at http://pubs.acs.org.

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